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state the law of multiple proportions states when two elements combine to form two or more compounds, the mass of one element that combines with a given mass of the other is in the ratio of small whole numbers State the two main ideas of Dalton's atomic theory that have remained unchanged since he first proposed his theory

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modern chemistry chapter 5 review. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. laurenrandall. Terms in this set (48) Cannizzaro. Developed a standard method for measuring ATOMIC MASS. This allowed chemists to search for periodic trends among elements. Dmitri Mendeleev. organized the elements according to increasing atomic mass and noticed that similar ...

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. In cathode-ray tubes, the cathode ray is emitted from the negative electrode, which is called the cathode . 2. The smallest unit of an element that can exist either alone

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or in molecules containing the same or different elements is the .atom 3. A positively ...

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