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valence electrons
does a neutral
magnesium (Mg)
atom have? 2. What
is the charge on a

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magnesium ion?
What does
magnesium have to
do to form such an
ion, and why does it
tend to do so? 3. How
many valence
electrons does a
single neutral chlorine
(Cl atom have? 4.

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Activity Series Use
with Chapter 9,
Section 9.2 1. For
each of the following
pairs of elements,
underline the one that
would replace the
other element in a
compound.

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Orbitals 1. What is the
shape of an s orbital?

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5.2 2. What is the
relationship between
the size of an s orbital
and the principal

energy level in which
it is found? 3. What is

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orbital? How many p
orbitals are there in a
sublevel? 4.

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Formation of Ions 1.

calcium and oxygen

2. Yes; each contains
equal numbers of

protons and electrons.

3. ionization energy 4.

positive; cation 5.

negative; anion 6. No;

it is not a stable octet
of electrons. 7.

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